

Metal–Organic Nanotube with Helical and Propeller-Chiral Motifs Composed of a C_{10} -Symmetric Double-Decker Nanoring

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Supporting Information

ABSTRACT: Coassembly of an achiral ferrocene-cored tetrapotic pyridyl ligand (FcL) with AgBF_4 in $\text{CH}_2\text{Cl}_2/\text{MeCN}$ (7:3 v/v) containing chiral Bu_4N^+ (+)- or (-)-menthylsulfate (MS^{*-}) results in the formation of an “optically active” metal–organic nanotube (FcNT) composed of a C_{10} -symmetric double-decker nanoring featuring 10 FcL units and 20 Ag^+ ions. The circular dichroism spectrum of FcNT along with its 2D X-ray diffraction (2D XRD) pattern indicates that the constituent metal–organic nanorings in FcNT stack one-handed helically on top of each other. A crystal structure of the dimeric double-decker model complex ($\text{Ag}_2(\text{FcL}')_2$) from a ditopic ferrocene ligand (FcL') and AgBF_4 allowed for confirming the binding of MS^{*-} onto the Ag^+ center of the complex. The results of detailed spectroscopic studies indicate that in its double-decker aromatic arrays, FcNT possibly possesses propeller-chiral twists in addition to the helically chiral structure, where the former is considerably more dynamic than the latter. Notably, both chiral structural motifs responded nonlinearly to an enantiomeric excess of MS^{*-} (majority rule) though with no stereochemical influence on one another.

Helices exhibit beautiful structural motifs that have been utilized for many historical architectures. In the mid 1970s, some synthetic polymers were found to adopt a helical geometry in solution,¹ and currently, a wide variety of helical polymers including noncovalent helical assemblies are known.^{2,3} Helical nanotubes are one of the most complicated classes of helical polymers since nanotubes are categorized into 1D architectures morphologically but composed of a rolled up 2D sheet. A successful synthetic strategy toward helical nanotubes would be to stack polygonal macrocycles on top of each other in a one-handed helical manner. Although successful examples of this strategy have been reported, they are only limited to the helical stacking of C_2 -, C_3 -, C_4 -, and C_6 -symmetric macrocycles.⁴ Herein, we report a new nanotubular assembly that is uniquely composed of a C_{10} -symmetric ultralarge coordination nanoring. Note that helical stacking of C_{10} -symmetric macrocycles is far more difficult

than that of lower-symmetry examples because the stacked nanorings must be twisted unidirectionally with a regular twisting angle of less than 18° (Figure 1b). Therefore, elaboration of the self-assembly is required.

In a previous study,⁵ we reported that a ferrocene-cored tetrapotic pyridyl ligand (FcL) and AgBF_4 coassemble in MeCN to afford a metal–organic nanotube (FcNT). Although this nanotubular coassembly occurs stochastically, FcNT is composed of a large number of decagonal nanorings with a diameter of 7.5 nm (Figure 1a), which is unusually large and has been out of range for direct synthesis using conventional coordination chemistry. In FcNT, the decagonal nanorings are stacked via π -electronic interactions. If they stack normally on top of each other, metallophilic (Ag–Ag) interactions among the stacked nanorings can also be expected. We initially thought that this normal stacking is the most likely candidate. However, a closer examination of the 2D X-ray diffraction (2D XRD) image of the magnetically oriented FcNTs indicated that a small helical twist may exist in the stacking geometry.⁶ If this helical twist can unidirectionally be developed over the entire nanotube, FcNT would be optically active. As described below, when the coassembly was carried out in the presence of (+)- or (-)-menthylsulfate (MS^{*-}), the resultant nanotube indeed displayed an optical activity.

Figure 2a shows a 2D XRD image of magnetically oriented FcNTs prepared under a 10 T magnetic field by slow solvent evaporation of an MeCN dispersion of a mixture of $\text{Bu}_4\text{N}^+\text{OTf}^-$ (8 mM) and FcNT, formed from FcL (4 mM) and AgBF_4 (8 mM) at 25°C . In addition to meridional and equatorial diffractions due to the intra- and intercolumnar orderings, respectively, of FcNT, two sets of four-split diffraction spots were observed at $q = 3.6$ (red arrows) and 6.9 nm^{-1} (blue arrows). Although the diffraction maxima were blurred, azimuthal plots at the q values in Figure 2c,e indicated the existence of two sets of four-split diffractions characteristic of the helical twists.⁶ Simulation of the 2D XRD pattern of FcNT (Figure 2a) using the HELIX⁷ program provided a simple atomistic model with a probable twisting angle of 4° (Figure S7).⁸ If this twisting results

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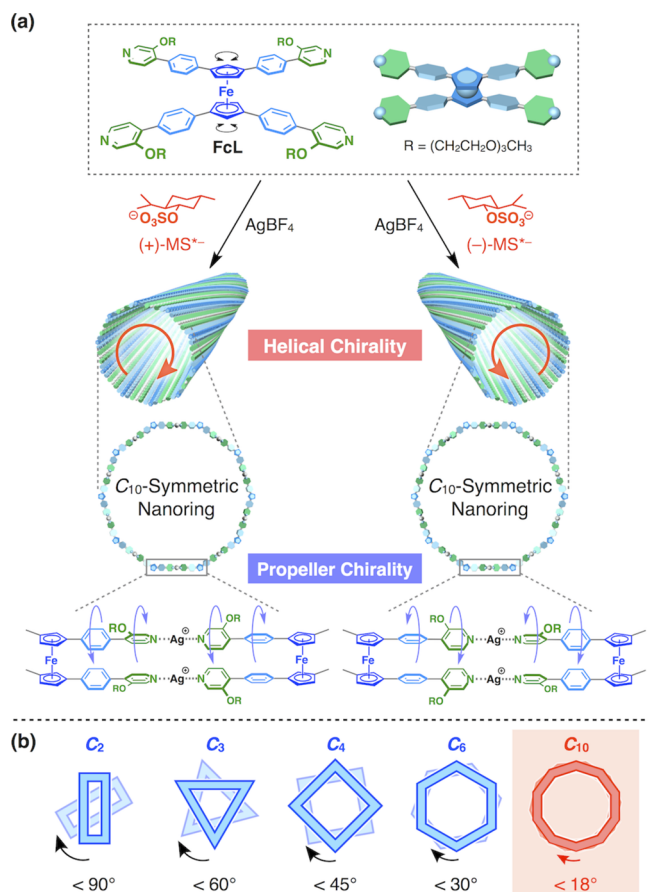


Figure 1. (a) Schematic representations of the coassembly of ferrocene-tetrapic ligand (FcL) and AgBF₄ in the presence of (+)- or (-)-methylsulfate (MS*⁻), and possible helical and propeller-chiral motifs in the resulting helical metal-organic nanotube (FcNT). (b) Upper-limit rotational angles for achieving the helical stacking of the C₂, C₃, C₄, C₆, and C₁₀-symmetric macrocycles.⁴

in a single-handed helix, a helical pitch of ca. 64 nm is expected. An analogous XRD pattern was observed for FcNT with (+)-Bu₄N⁺MS*⁻ (Figure 2b,d,f), indicating that FcNT is intrinsically composed of helical twists, regardless of whether FcNT carries chiral counterions or not.

Prior to detailed studies of FcNT, we attempted to synthesize a dimeric complex (Ag₂(FcL')₂) as a model for the double-decker constituent of FcNT. Therefore, a ditopic ferrocene ligand (FcL') bearing only two pyridyl groups (4 mM) was coassembled with AgBF₄ (4 mM) in the presence of (+)-Bu₄N⁺MS*⁻ (4 mM) in THF/MeCN (99.5:0.5 v/v) (Figure 3a).⁸ When cyclohexane vapor was allowed to diffuse slowly at 25 °C into this coassembling mixture, a single crystal appropriate for X-ray crystallography was formed. The crystal adopted a chiral space group of P2₁2₁2₁ with a composition of Ag₂(FcL')₂(BF₄)₂((+)-MS*⁻)(THF), as determined by elemental analysis.⁸ The ORTEP diagram in Figure 3b indicates that two FcL' molecules adopt an eclipsed geometry and are ditopically connected by Ag⁺ ions to form a double-decker structure. This structure is consistent with the building unit for FcNT proposed in our previous study.⁵ Notably, a bulky (+)-MS*⁻ anion is located on one side of Ag₂(FcL')₂ and its sulfate group bridges two Ag⁺ atoms. In addition, a BF₄⁻ anion is located on the other side of the double-decker structure. The crystal packing diagram of

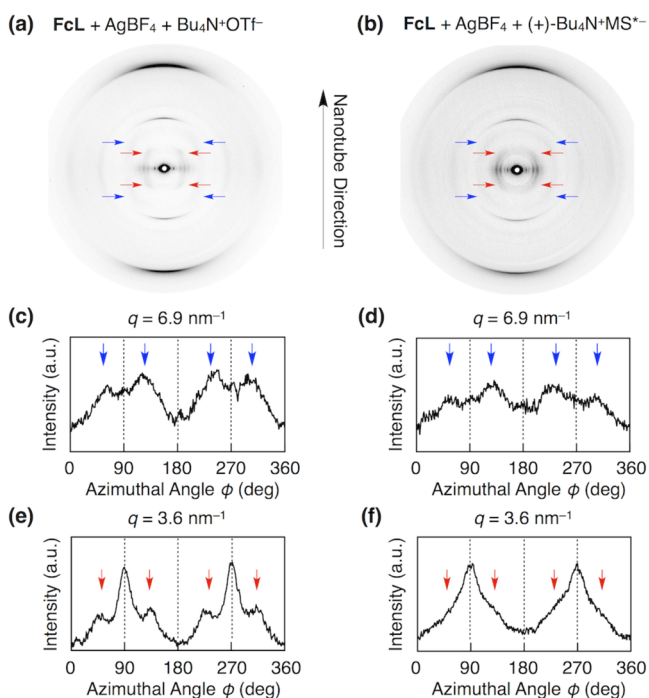


Figure 2. Two-dimensional X-ray diffraction (2D XRD) patterns of film samples of magnetically oriented FcNTs carrying (a) OTf⁻ prepared from FcL (4 mM), AgBF₄ (8 mM), and Bu₄N⁺OTf⁻ (8 mM), and (b) MS*⁻ prepared from FcL (2 mM), AgBF₄ (4 mM), and (+)-Bu₄N⁺MS*⁻ (4 mM) in MeCN. Azimuthal plots for the (a,b) 2D XRD intensities at $q =$ (c,d) 6.9 and (e,f) 3.6 nm⁻¹ as a function of the azimuthal angle (ϕ). Red and blue arrows indicate that the diffraction spots split into four-spot patterns.

Ag₂(FcL')₂ (Figure 3c) indicates that Ag₂(FcL')₂ stacks on top of each other due to π -electronic interactions.

As a typical example of nanotubular coassembly, a CH₂Cl₂/MeCN (7:3 v/v) solution consisting of a mixture of FcL (100 μ M), AgBF₄ (200 μ M), and (+)-Bu₄N⁺MS*⁻ (200 μ M) was allowed to stand at 25 °C for 12 h without stirring. Then, a transparent orange dispersion resulted. As shown in Figure 4a, this dispersion exhibited an intense absorption band at 335 nm (solid curve), which was red-shifted from that of nonassembled FcL at 307 nm (broken curve) due to the π - π^* transition of its π -conjugated arms. Transmission electron microscopy (TEM; Figure 4b) and atomic force microscopy (AFM; Figure S6) studies of its air-dried residue indicated the formation of hollow cylinders with a uniform diameter of 7.3 nm. When Bu₄N⁺MS*⁻ was mixed with FcNT ([MS*⁻]/[Ag⁺] = 1.0) in CD₃CN, ¹H NMR signals due to MS*⁻ disappeared, but those due to Bu₄N⁺ remained unchanged (Figure S5), indicating that MS*⁻ is tightly bound to Ag⁺ (see also Figure 3b).⁹ FcNT, which formed in the presence of (+)- or (-)-MS*⁻, was optically active and exhibited Cotton effects at 325 and 343 nm in its circular dichroism (CD) spectrum (Figure 4c). This result is exciting because FcNT is composed of a C₁₀-symmetric nanoring and can be optically active only when the twisting angle of the stacked nanorings is less than 18°. The CD spectrum of FcNT with (+)-MS*⁻ was a mirror-image of that with (-)-MS*⁻.

Next, the dynamic nature of this unique helical nanotube was investigated. First, FcNT was prepared in the presence of (+)-Bu₄N⁺MS*⁻ (200 μ M) in CH₂Cl₂/MeCN (7:3 v/v) under ordinary conditions, and an excess amount of (-)-Bu₄N⁺MS*⁻ (600 μ M) was added to the resulting dispersion. Interestingly,

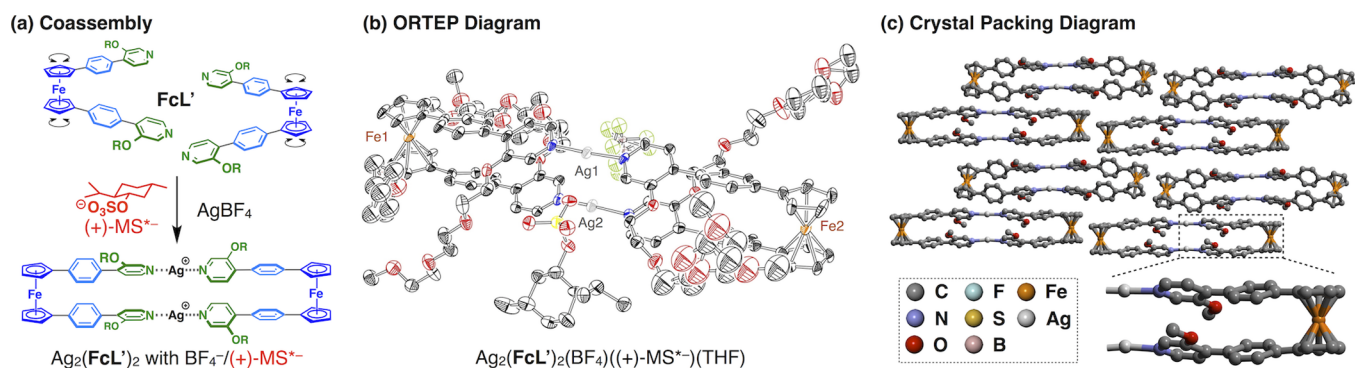


Figure 3. (a) Schematic representation of the coassembly of FcL' with AgBF_4 in the presence of $(\text{+})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$. (b) An ORTEP diagram of the $\text{Ag}_2(\text{FcL}')_2(\text{BF}_4)((\text{+})\text{-MS}^{*-})(\text{THF})$ dimeric complex with a probability level of 30%. Three of the four triethylene glycol (TEG) chains, BF_4^- , and a THF molecule are disordered. (c) A crystal packing diagram of $\text{Ag}_2(\text{FcL}')_2(\text{BF}_4)((\text{+})\text{-MS}^{*-})(\text{THF})$ with a magnified image of the aromatic rings of FcL' . The TEG chains, BF_4^- , MS^{*-} , and a THF molecule are omitted for clarity.

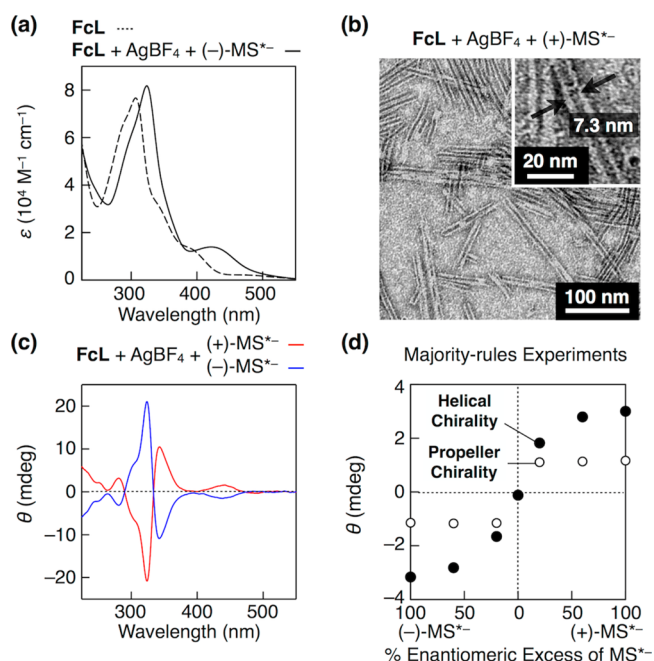


Figure 4. Coassembly of FcL ($100 \mu\text{M}$) with AgBF_4 ($200 \mu\text{M}$) at 25°C in $\text{CH}_2\text{Cl}_2/\text{MeCN}$ (7:3 v/v) containing $(\text{+})\text{-}$ or $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ ($200 \mu\text{M}$). (a) Electronic absorption spectra of FcNT prepared in the presence of $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ (solid curve) and FcL alone ($100 \mu\text{M}$; broken curve) as a reference in a 1.0 mm-thick quartz cell. (b) A TEM image of an air-dried sample of FcNT prepared in the presence of $(\text{+})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ and stained with uranyl acetate. Inset shows a magnified TEM image. (c) CD spectra of FcNT assembled in the presence of $(\text{+})\text{-}$ (red curve) and $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ (blue curve) in a 1.0 mm-thick quartz cell. (d) Changes in the θ values at 278 (O) and 352 nm (●) as a function of the % enantiomeric excess (%ee) of MS^{*-} .

minor CD bands corresponding to FcNT at 240–290 nm underwent inversion, whereas the residual major bands remained virtually intact (Figure 5a), suggesting that FcNT contains two chiral structural motifs with different dynamic properties. The minor CD bands only disappeared in 0.3 h (Figure 5c). Therefore, a split Cotton effect, which is characteristic of helical motifs,¹⁰ was observed with positively and negatively signed CD bands at 340 and 322 nm, respectively (Figure 5c). As discussed later, this CD activity is long-lived. The dynamic CD components were extracted (Figure 5d) by subtracting this particular spectrum from the CD spectra in Figure 5a. We assume

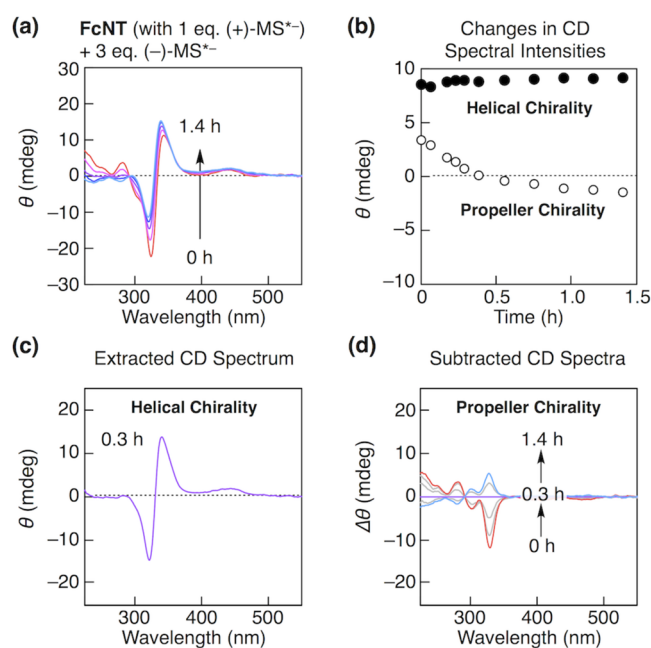


Figure 5. Addition of $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ to a dispersion of FcNT ($[(\text{-})\text{-MS}^{*-}]/[\text{Ag}^+] = 3.0$) in $\text{CH}_2\text{Cl}_2/\text{MeCN}$ (7:3 v/v), prepared by the coassembly of FcL ($100 \mu\text{M}$) with AgBF_4 ($200 \mu\text{M}$) at 25°C in the presence of $(\text{+})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$ ($200 \mu\text{M}$). (a) Time-dependent CD spectral change after the addition of $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$. (b) Plots of the CD spectral intensities at 278 (O) and 352 nm (●) due to the propeller-chiral and helical geometries, respectively. (c) A CD spectrum extracted from panel a, which was observed in 0.3 h upon addition of $(\text{-})\text{-Bu}_4\text{N}^+\text{MS}^{*-}$. (d) Calculated CD spectra obtained by subtraction of the CD spectrum in panel c from those in panel a.

that the domino-like condensed aromatic arrays in FcNT adopt a propeller-chiral geometry (Figure 1a), which results in these dynamic CD bands. Based on a closer examination of the structure of $\text{Ag}_2(\text{FcL}')_2$ in Figure 3c, the phenyl and pyridyl rings in FcL' are tilted in opposite directions by $+17$ and -21° (or -17 and $+21^\circ$), respectively, relative to the Cp rings in the proximal ferrocene unit. Such propeller-chiral structures are well-known for *ortho*-substituted biaryls.¹¹ Time-dependent DFT calculations based on a dimeric model complex with an energy-minimized geometry (Figure S8) provided a simulated CD spectrum (Figure S9), which is consistent with the one observed for $\text{Ag}_2(\text{FcL}')_2$ in THF at -80°C .⁸ As described above, by comparison with the CD spectral change for the propeller-chiral

structure (Figure 5b, open circles) in FcNT, the change due to the helical structure was much more sluggish (Figure 5b, filled circles), where only 14% CD attenuation at 352 nm occurred in 6.9 h.

In Figure 4d, (+)- and (-)-Bu₄N⁺MS^{*-}, mixed at varying mole ratios, were employed for the formation of FcNT, where the CD intensities at 278 (open circles; propeller chirality) and 352 nm (filled circles; helical chirality) both changed sigmoidally with the % enantiomeric excess (%ee) of MS^{*-}. Therefore, the two different chiral motifs in FcNT (Figure 1a) follow the majority rule.¹² Note that the aromatic units in FcNT, adopting a propeller-chiral geometry, are sterically congested, so that their conformational changes may not be free from one another but synchronous like a “domino”.¹³ The sigmoidal response of the helical chirality to the %ee of MS^{*-} may also be reasonable, considering that each nanoring can accommodate 20 chiral anions, which must determine the twisting direction of the nanoring based on their majority voting. Of interest, the propeller and helical geometries in FcNT show different nonlinear responses to the %ee of MS^{*-} (Figure 4d). So, they do not influence on one another.

In conclusion, the nanotubular coassembly between achiral tetrapropyl pyridyl ligand FcL and AgBF₄ in the presence of a chiral Bu₄N⁺ salt of (+)- or (-)-menthylsulfate (MS^{*-}) (Figure 1a) resulted in the formation of a metal–organic nanotube (FcNT) in an optically active form due to the successful stereochemical bias of the helical twist in the stacking geometry of the constituent nanorings in either a left- or right-handed manner. This achievement is noteworthy because the nanoring is highly symmetric (C₁₀-symmetric; Figure 1b) and needs fine-tuning in its stacking geometry to generate a one-handed helical twist. In addition to this helical chirality, FcNT possesses the propeller chirality (Figure 1a). Although both chiral motifs exhibit a nonlinear response to the %ee of the chiral auxiliary (MS^{*-}), their dynamic natures are different. The propeller-chiral motif is dynamic, whereas the helical motif is considerably less dynamic. Application of this unique chiral metal–organic nanotube for enantioselective catalysis and separation provides important areas for future investigation.

■ ASSOCIATED CONTENT

Supporting Information

Details of synthesis and characterization of FcL' and (+)-/(-)-Bu₄N⁺MS^{*-}, analytical data by NMR, CD spectroscopy, atomic force microscopy, 2D XRD, and X-ray crystallography; and results of DFT calculations. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b04386.

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Notes

The authors declare no competing financial interest.

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■ NOTE ADDED AFTER ASAP PUBLICATION

Figure 4 was corrected on June 16, 2015.